Periodic graphics

A collaboration between C&EN and Andy Brunning, author of the popular graphics blog Compound Interest



To see more of Brunning's work, go to compoundchem.com. To see all of C&EN's Periodic Graphics, visit http://cenm.ag/ periodicgraphics.

THE CHEMISTRY OF CANNING

To preserve fruits and vegetables, you must first sterilize them and then seal them in airtight containers. In this graphic, we take a look at some of the key parts of the process.



Two types of foods are destined for canning. High-acid foods have a high enough acidity to block the growth of bacteria. Low-acid foods need to have acid added (often from lemon juice or vinegar), or they must be sterilized at a higher temperature (see right).

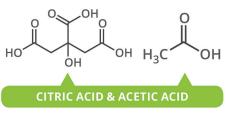
HIGH-ACID FOODS PH < 4.6

Fruits and foods with added acid (e.g., pickles)





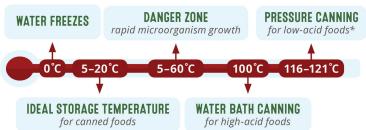
Red meat, poultry, seafood, and fresh vegetables (tomatoes can have pH > 4.6 in some cases)



Both can be used to lower the pH of low-acid foods to ensure it is below pH 4.6



THE IMPORTANCE OF TEMPERATURE

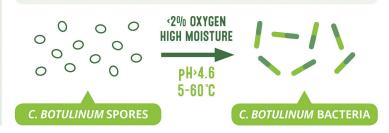


* Pressure cooker at 10 psi (116 °C) and 15 psi (121 °C) needed to reach these temperatures

Heating foods to the correct temperature is important because it kills microbes that might otherwise grow under some storage conditions. In the case of low-acid foods, you must use a pressure cooker to reach the required temperature. The temperature at which jars are stored can also affect the quality of canned produce.

BOTULINUM SPORES AND CANNING

Botulinum spores reside on most fresh foods. Though they're usually harmless, under the right conditions, they can germinate into bacteria that make deadly botulinum toxin. Acidity and temperature are key to preventing toxin production in canned goods.



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