

Periodic Graphics

A collaboration between C&EN and
Andy Brunning, author of the popular
graphics blog *Compound Interest*

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online

To see more of
Bunning's work, go to
compoundchem.com.
To see all of C&EN's
Periodic Graphics,
visit [cenm.ag/
periodicgraphics](http://cenm.ag/periodicgraphics).

COMBATING UNDERARM ODOR

What makes our armpits smell when we sweat? And how do deodorants and antiperspirants fight the odor? Here we take a look at the chemistry involved.

WHAT CAUSES UNDERARM ODOR?

Microbes that live on our skin transform odorless underarm secretions such as amino acids into stinky molecules. Our resident underarm microbes consist mainly of four groups of bacteria, but corynebacteria are the primary cause of underarm odor.



CORYNEBACTERIUM



MICROCOCCUS



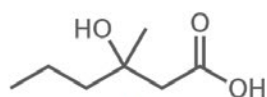
CUTIBACTERIUM



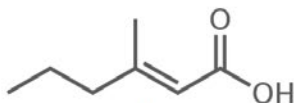
STAPHYLOCOCCUS

MALODOROUS MOLECULES

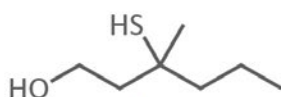
Short- and medium-chain volatile fatty acids are the main contributors to underarm odor, along with a small number of thiols.



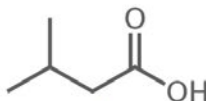
3-HYDROXY-3-METHYLHEXANOIC ACID



(E)-3-METHYL-2-HEXENOIC ACID



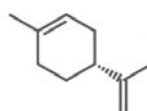
3-METHYL-3-MERCAPTOHEXAN-1-OL



ISOVALERIC ACID

DEODORANTS AND ANTIPERSPIRANTS

Deodorants and antiperspirants include pleasant-smelling compounds to mask the stinky ones. Some common fragrance molecules include limonene, which has a citrus scent, and pinene, which has a pine smell.



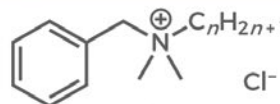
LIMONENE



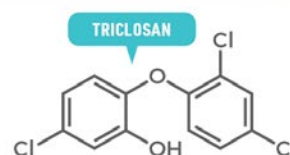
α -PINENE



Deodorants and antiperspirants can also include antimicrobial compounds that kill bacteria that produce malodorous molecules. Common antimicrobials are benzalkonium chloride and triclosan.

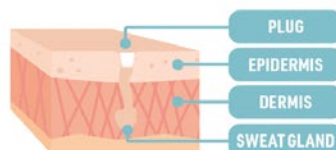


BENZALKONIUM CHLORIDE



TRICLOSAN

Antiperspirants rely on aluminum-based compounds that form polymeric plugs to block perspiration from escaping sweat glands. With less perspiration around, underarm bacteria make fewer malodorous molecules.



ALUMINUM CHLOROHYDRATE